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SEDIMENTARY PETROLOGY (ii)

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- <u>Boggs</u>, S.Jr. (2009): <u>Petrology of Sedimentary</u>
 <u>Rocks</u>, 2 nd ed., Cambridge Univ. Press, 600p.
- Selley, R.C. (2000): <u>Applied Sedimentology</u>, 2 nd ed., Academic Press, 543p.

Sedimentary Petrology

Sedimentary Petrology: is that particular

branch of study concerned especially with

the composition, characteristics, and origin

of sediments and sedimentary rocks

(Boggs, 2009).

What are Sedimentary Rocks

- Sedimentary rocks are formed at low temperatures and pressures at the surface of Earth owing to deposition by water, wind or ice.
- Sedimentary rocks cover about 80% of the total land area of Earth.
- They make up **about 11%** of the volume of the Earth's crust (Ronov, 1983).
- They also cover most of the ocean's floor above a basement of volcanic rocks.

What are Sedimentary Rocks

- Sedimentary rocks are characterized by the presence of layers and by distinctive textures, structures, mineral composition and fossil content.
- They provide us available clues to evolution of Earth's landscapes and life forms through geologic time.
- In addition, many sedimentary rocks contain important economic products such as water, petroleum, natural gas, minerals, metallic ores, uranium, coal, iron, salts, etc...

Source: Petrology of Sedimentary Rocks, by 5. Boggs, Jr. (2009)

What are Sedimentary Rocks

Sedimentary rocks are composed of either:

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 particles derived from pre-existing rocks (Terrigenous or Clastic sedimentary rocks)

* rocks precipitated from water by chemical or biochemical processes (Autochtonous

Sed, rocks).

Source: Petrology of Sedimentary Rocks, by S. Boggs, Jr. (2009)

Sedimentary Rocks





Take resources from them





How do sedimentary rocks form?

- Weathering & Erosion (Mechanical breakdown, Chemical breakdown)
- Transport (Water, Air, Glaciers)
- **Deposition & Lithification**
 - Deposited in layers (strata)
 - Layers are progressively younger towards the top of a succession.

Terrestrial processes

- Rivers, Lakes, Deltas, Fans...

Precipitation

- Marine/ groundwater processes
- Sea, Lakes, Cave systems...

Recognizing sedimentary rocks

- Stratification visible (Bedding Lamination)
- Fossils present
- Grains or pebbles visible
- Sedimentary minerals present
- Sedimentary structures visible on bedding planes

Components of Sedimentary Rocks

Four Fundamental kinds of components:

• **Terrigenous siliciclastic particles:**

(rock fragments - quartz - feldspars – micas -clay minerals & iron oxides).

- Chemical/biochemical components: (calcite – gypsum – apatite <u>as well as</u> calcareous and siliceous tests or shells of organisms)
- Carbonaceous components :

(carbonized residues of terrestrial plants (Humic materials) and marine plants & animals, Petroleum bitumens)

 Authigenic (secondary) components : (quartz – feldspars – clay minerals – calcite – gypsum – barite - hematite) Classification of Sedimentary Rocks

- Terrigenous Rocks (Siliciclastic Rocks):
 Conglomerates, Sandstones, Mudrocks.
- Marine Rocks (Biogenic, Biochemical Rocks):
 Limestones & Dolomites.
- Others:
 - Evaporites (chemical Rocks).
 - Siliceous rocks (Cherts) (chemical Rocks).
 - Ironstones (chemical Rocks).
 - Phosphates or Phosphorites.
 - Organics (Coals and Oil Shales).
 - Volcaniclastic Rocks (Pyroclastic Rocks).

Classification of Sedimentary Rocks

osition	Group name	Particle size	Principal constituents	Main rock types	
sidues <50% Terrigenous siliciclastic grains	or siliciclastic rocks	>2 mm	Rock tragments	Conglomerates and breccias	
		1/16-2 mm	Silicate minerals and rock fragments	Sandstones	
		<1/16 mm	Silicate minerals	Shales (mudrocks)	
<~15% Carbonaceous re >50% Chemical/biochemical constituents	Chemical/biochemical rocks	Variable	Carbonate minerals, grains; skeletal fragments	Carbonate rocks (limestones and dolomites)	
			Evaporite minerals (sulfates, chlorides)	Evaporites (rock salt, gypsum, anhydrite)	
			Chalcedony, opal, siliceous skeletal remains	Siliceous rocks (cherts and related rocks)	
			Ferruginous minerals	Ironstones and iron-formations	
			Phosphate minerals	Phosphorites	
>-15% Carbonaceous residues		Variable	Siliciclastic or chemical- biochemical constituents: carbonaceous residues	Oil shales Impure coals	
			Carbonaceous residues	Humic coals Cannel coals	
	 >50% Chemical/biochemical <50% Terrigenous siliciclastic grains 	residues >50% Chemical/biochemical uoitical >50% Chemical/biochemical <50% Terrigenous	Carbonaceous rocks Carbonaceous rocks Siliciclastic clastic or siliciclastic or siliciclastic rocks Natiable Variable Variable Variable Variable Variable Variable	Osition Group name Particle size Principal constituents >2 mm Rock Iragments >2 mm Silicate minerals and rock fragments 00% 1/16-2 mm 1/16-2 mm Silicate minerals 00% 1/16 mm 1/16 Silicate minerals 00% 1/16 mm 100% Silicate minerals 100% Silicate minerals 100% Silicate minerals 100% Silicate minerals 110% Silicate minerals 110% Silicate minerals 1116 Silicate minerals	

Source: Petrology of Sedimentary Rocks, by S. Boggs, Jr. (2009)

Abundance of Sedimentary Rocks

Ronov (1983) suggest the followings:

- Shales makes up ~50% of Sed. Rocks on the continents
- Sandstones & conglomerates makes up ~24% of Sed. Rocks on the continents
- Carbonates makes up ~24% of Sed. Rocks on the continents
- Evaporites makes up $\sim 1\%$ of Sed. Rocks on the continents
- Cherts makes up $\sim 1\%$ of Sed. Rocks on the continents
- Phosphorites and carbonaceous sedimentary rocks are quit small.

- *Five main genetic classes of sediment* can be recognized *according to Hatch et al. (1971):* chemical, organic, residual, terrigenous, and pyroclastic .
- The chemical sediments: are those that form by direct precipitation in a subaqueous environment.
 Examples include evaporites such as gypsum and rock salt, as well as tufa and some lime muds.

2) The organic sediments: are those composed of organic matter of both animal and vegetal origin.
 Examples include skeletal limestones and coal.

3) The residual sediments: are those left in

place after weathering.

Examples include the *laterites and bauxites.*

4) The terrigenous sediments: are those whose particles were originally derived from the earth, and include the mudrocks, siliciclastic sands, and conglomerates.

4) Pyroclastic sediments: are the product of volcanic activity. Examples include ashes, tufs, volcaniclastic sands, and agglomerates.

• These five main genetic classes of sedimentary rocks can be divided into 2000

separate types:

- The allochthonous and
- the autochthonous deposits.

- The allochthonous sediments are: those that are transported into the environment in which they are deposited.
- They termed Extrabasinal Sediments
- They include the terrigenous and pyroclastic

classes, with rare reworked carbonates.

- The autochthonous sediments are: those that form within the environment in which they are deposited.
- They termed Intrabasinal Sediments.
- They include the chemical, organic, and residual classes.

ALLOCHTHONOUS SEDIMENTS (EXTRABASINAL SEDIMENTS)

ALLOCHTHONOUS SEDIMENTS

- The allochthonous sediments are classified to four types:
- Terrigenous sediments:
 - Mudrocks
 - Siliciclastic sands
 - Conglomerates.

Pyroclastic sediments (ashes, tuffs, volcani-clastic sands & agglomerates).

ALLOCHTHONOUS SEDIMENTS



Classification of unconsolidated sediments

on a basis of grain size (Shepard, 1954)





Siliciclastic Sedimentary Rocks

Siliciclastic (terrigenous) rocks are classified based on the grain size into three divisions:

- Gravels & Conglomerates: consist of clasts >2mm in diameter.
- **2. Sand & Sandstones:** the particles range in diameter between 2mm and 63 μm.
- *3. Muds & Mudstones (including clay and silt):* made up of particles < 63 μm.

Wentworth scale of grain size classification

Millimeters (mm)	Micrometers (µm)	Phi (ø)	Wentworth size class	Rock type Conglomerate/ Breccia
4096 256 64 4		-12.0 -8.0 — -6.0 — -2.0 —	Boulder Cobble Pebble Granule	
2.00 1.00 1/2 0.50 1/4 0.25 1/8 0.125		- 0.0 — - 1.0 — - 2.0 — - 3.0 —	Very coarse sand Coarse sand Medium sand Fine sand Very fine sand	Sandstone
1/16 0.0625 1/32 0.0310 1/64 0.0156 1/128 0.0078 1/256 0.0025	63 63 $ -$ 31 $ -$	4.0 — 5.0 — 6.0 — 7.0 —	Coarse silt Medium silt Fine silt Very fine silt	Siltstone
0.0039	6 0.06	14.0	Clay	Claystone

Clastic Sedimentary Rocks								
Texture (g	grain size)	Sediment Name	Rock Name					
Coarse (over 2 mm)		Gravel (rounded fragments)	Conglomerate					
		Gravel (angular fragments)	Breccia					
Medium (1/16 to 2 mm)		Sand	Sandstone					
Fine (1/16 to 1/256 mm)		Mud	Siltstone					
Very Fine (less than 1/256)		Mud	Shale					



- The term "mud" in Recent deposits, refer to sediments that are "wet clays" with a certain amount of silt and sand.
- In lithified equivalents, are termed "mudstones".
- In Wentworth scale:

Mud comprise: clay sediments (particles: < 0.004 mm)

& Silt sediments (particles: 0.004 mm – 0.063 mm).

Their lithified equivalents are:

Claystones & siltstones.



Shale is another term
applied to fine-grained
sediments that have partings
called *fissility*.

This is due to traces of mica aligned on laminae at the time of deposition.







Clays are deposited with a primary water-saturated porosity of up to 80%.

 Most of this porosity is quickly lost, first by dewatering and later by compaction.



Major constituents of mudrocks

- **Major constituents of mudrocks include:**
- clay minerals, detrital grains, organic matter and carbonates.
- **Clay Minerals:** kaolin group, smectite group, illite group, chlorite & glauconite.
- **Detrital grains:** quartz, mica and heavy minerals.
- Organic matter: is chemically very complex & composed of kerogen, asphalt, crude oil and natural gas.

Nomenclature & composition of mudrocks

- Mudrocks can be named with reference to an endmember triangle apices represent:
- Organic Matter,
- Pure Lime (carbonates)
- Pure Clay Minerals.



Nomenclature & composition of mudrocks

- Mudrocks composed largely of admixtures of clay minerals termed orthoclaystones.
- With increasing lime content, claystones grade into marls and into micrites (calcilutites) which are pure lime mudrocks.



Nomenclature & composition of mudrocks

- Mudrocks with traces of organic matter are referred to as *organic claystones (organic shales).*
- With increasing organic content, organic claystones (organic shales) grade into oil claystones (oil shales) and thence into the dominantly carbonaceous sapropelites and coals.
- The organic-rich mudrocks are thought to be the *source rocks* from which liquid hydrocarbons are generated.



Nomenclature & composition of mudrocks

- Mudrock needs to contain over 1.5% organic carbon to be a significant source rock.
- The type of kerogen determine the type of hydrocarbon which may be generated:
 - Terrestrial humic kerogen tends to be gas prone.
 - Algal kerogen tends to be oil prone.
 - Mixed kerogen can generate both oil & gas.
- Temperature is also important for oil & gas generation;
 - Oil generation takes place between 60-120 °C.
 - Gas generation occurs between 120-220 °C.

ORGANIC MATTER

The organic matter in sediments is of four types:
 Kerogen, asphalt, crude oil & natural gas. Kerogen: is a dark greyish-black amorphous solid , present in varying amounts in mudrocks, when pure it is coal.

- The exact molecular structure of kerogen is not well known.
- It includes hydrocarbon compounds which are insoluble in normal petroleum solvents.

Kerogen is the major consituent of the organic-rich mudrocks.
ORGANIC MATTER

- Asphalt or bitumen, is similar to kerogen, but it is soluble in normal petroleum solvent.
- **Asphalt** occurs both in infilling sediment pores and fractures.
- **Crude Oil** comprises hydrocarbons which are liquid at normal temperatures and pressure.
- Crude Oils are generally consist of varying proportions of four main groups, The paraffins, aromatics, naphthenes, and asphalts.
- **Crude Oil** occurs in pore spaces of many rocks in favourable circumstances.

ORGANIC MATTER

Table 8.2 Properties and composition of the main groups of organic hydrocarbons						
		Average composition (% weight)				
Organic matter	Properties	С	H ₂	$S + N + O_2$, etc.		
Kerogen Asphalt	Solid at surface temperatures and pressures. Insoluble in normal petroleum solvents Solid or plastic at surface temperatures and pressures.	75	10	15		
Canada ail	Soluble in normal petroleum solvents	83	10	7		
Crude on	and pressures	85	13	2		
Natural gas	Gaseous at surface temperatures and pressures	70	20	10		

ORTHOCLAYSTONES



ORTHOCLAYSTONES

- The orthoclaystones composed *mainly of the clay minerals groups*.
- There are five principal groups of clay minerals: *illites, smectites (montmorillonites), kaolins, Chlorites and glauconite.*
- The last two differ from the other clay minerals in mode of formation, but show similarities in composition & atomic structure.
- All five mineral groups are *hydrous alumino-silicates*.
- All clay minerals are crystalline except glauconite is amorphous.

CLAY MINERALS

 The clays are composed of different arrangement of tetrahedra and octahedra layers.





Linkage of Terahedra & Octahedra



Kaolin group (1:1 Structure)

 The kaolin group of clay minerals includes *kaolinite, dickite,* nacrite, halloysite and allophane.



 The differences between these minorals are both Si 0.72 nm

0

Al • Si







AI



Kaolin group (1:1 Structure)

- Kaolinite is the simplest clay mineral in structure and the purest in composition [Al₂Si₂O₅(OH)₄].
- It is formed from feldspars by *hydrothermal alteration & superficial weathering.*
- Kaolinite is the common detrital clay mineral in sediments derived from granitic and gneissose sources.
- Kaolinite is an important authigenic cement in some sandstones.



Kaolin group (1:1 Structure)







- In certain circumstances kaolin sedimentation is sufficiently abundant to form *a pure kaolin claystone*.
- The rock type is variously termed *pipe clay, china* clay & fire clay.
- **Tonstein** is another distinctive rock type, composed largely of kaolinite.
- Kaolinite is extensively used in ceramic, paper-making and pharmaceutical industries.
- Pure kaolinite rocks are non-marine in origin because kaolinite quickly transforms to more complex clays in the presence of sea water.

ILLITE GROUP (2:1 Structure)

- The illite clays, sometimes termed the *hydromicas*.
- They composed of three-layer aluminosilicates with up to 8% K_2O .
- [K_{0.8-0.9}(Al,Fe,Mg)₂(Si,Al)₄O₁₀(OH)₂]
- Potassium may either be present due to incomplete degradation of K-feldspar to kaolinite, or to diagenesis of kaolinite
 Within marine environment.





ILLITE GROUP (2:1 Structure)

- Illite is less obvious than kaolinite because it is seldom present in crystals.
- Under SEM, illite crystals are smaller and less well developed than those of kaolinite.





SMECTITE GROUP

(2:1 Structure)

- These are three-layer types which have unusual property of *expanding & contracting* to absorb or lose water.
- *Montmorillonite* is the chief example of smectite group.
- Mont. Can contain up to 20% water as well as Ca & Mg ions.



SMECTITE GROUP

(2:1 Structure)

- Mudrocks composed largely of smectite clays termed *Bentonites*.
- Bentonites are formed by the alteration of volcanic ash in situ.
- This may occur in both marine & continental environments.
- They composed of fragments of detrital glass, with microscopic grains of quartz, mica, feldspars & heavy minerals.





Bentonites

- Bentonites are a major constiuent of circulating mud systems used in rotary drilling.
- Conversely, montmorillonite is an unfortunate clay mineral as a matrix in an oil reservoir.
- During production, water entering the reservoir may cause the clay matrix to expand and thus destroy permeability.



CHLORITES (2:1:1 STRUCTURE)

The chlorites are mixed-layer lattice clays with up to 9% FeO and 30% MgO.

> * The overall chemical composition of Chlorite minerals is:

> > $(Mg, Fe, AI)_6 (Si, AI)_4 O_{10} (OH)_8$

The common chlorite minerals are: <u>Clinochlore</u> (Mg-rich chlorite), <u>Chamosite</u> (Fe-rich chlorite).



CHLORITES (2:1:1 STRUCTURE)

- Chlorites have quite a variable chemical compositions due to a lot of the Al is normally replaced by Mg and Fe.
- Other metals including Zn, Mn and Ni
 can also replace Al .
- Chlorites occur as an alteration product of micas.
- Chlorite is a common accessory detrital mineral in immature sands and mudrocks.
- It is characteristic constituent of the microcrystalline matrix of greywackes.



GLAUCONITE (2:1 Structure)

- The term glauconite is used in two ways:
- It is applied to pretty, rounded green grains commonly seen in marine sediments, and



- The term is applied to a particular mineral.
- Analyses of glauconites show them to contain a mixture of clays whose lattices are in various

GLAUCONITE

- Glauconite proper is a three-layer clay mineral containing *Mg, Fe & K*.
- The chemical composition: (K,Na)(Fe,Al,Mg)₂(Si,Al)₄O₁₀(OH)₂
- It occurs in dark green amorphous grains seldom larger than fine-sand grade.

It is found in mudrocks & sandstones.





GLAUCONITE (2:1 STRUCTURE) © Geochemical evidence suggests that glauconite formation occurs in seawater at

low temperatures *in an environment which is neither strongly oxidizing nor reducing.*

- Optimum depth for glauconite genesis appears to be between 50 and 1000 m below sea level.
- Glauconite is abundant only in sea-floor areas that are isolated from large supplies of land-derived sediment.

Glauconite forms by three principal processes:

- 1. Replacement of the fecal pellets of bottom-dwelling organisms and by infilling foraminiferal tests & larger shells.;
- 2. Modification of particles of illitic and biotitic clays by seawater;
- 3. Direct precipitation from seawater.



INTRODUCTION

- About 30% of the land's sedimentary cover is made of terrigenous sand and sandstone.
- Sandstones make up nearly *one-quarter* of the sedimentary rocks in the geologic record.
- They form under a wide range of depositional conditions in a variety of depositional environments.
- Because they are often *highly porous*, sandstones are frequently *major aquifers and petroleum reservoirs*.
- They are more uniform in stratigraphy and petrophysical character than carbonate rocks.



Classification of sandstones according to Folk (1968). *Q: Quartz, F: Feldspars, L:*



Classification of sandstones according to Folk (1980). *Q: Quartz, F: Feldspars, R:*



Classification of sandstones after Pettijohn *et al.* (1987).

• One of the most fruitful concepts on which

sandstone nomenclature is based on the

idea of maturity (maturation).

• The maturation of a sand takes place in two ways:

It matures *chemically,* and

it matures *physically*.

Chemical Maturity (Mineralogical M.):

- Sediments form from the weathering of mineralogically complex source rocks.
- Throughout weathering and transportation relatively

unstable minerals are destroyed and chemically

stable minerals thus increase proportionally.

- Quartz is the most abundant stable mineral & feldspar is a common example of an unstable mineral.
- An index of the chemical maturity of a rock might be: the ratio of quartz to feldspar.
- As sediments are reworked through two or more cycles of sediment, they thus *tend to be mature* to pure quartz sands.

Intensity of weathering



Minerals Remaining in Clastic Sediments Derived from an Average Granite Outcrop Under Varying Intensities of Weathering

INTENSITY OF WEATHERING

Low	Medium	High
Quartz	Quartz	Quartz
Feldspar	Feldspar	Clay minerals
Mica	Mica	
Pyroxene	Clay minerals	
Amphibole		

Resistance of Minerals to Weathering



Chemical Stability of Minerals during Weathering

Stability of minerals	Rate of weathering		
MOST STABLE	Slowest		
Iron oxide (hematite)			
Aluminum hydroxides (gibbsite)			
Quartz			
Clay minerals			
Muscovite mica			
Potassium feldspar			
Biotite			
Albite (Na-rich feldspar)			
Amphiboles			
Pyroxene			
Anorthite (Ca-rich feldspar)			
Olivine			
Calcite			
Halite			
LEAST STABLE	Fastest		

Nomenclature and Classification of Sandstones <u>Physical Maturity (Textural M.)</u>:

- Physical maturity describes the textural changes that a sediment undergoes from the time it is weathered until it is deposited.
- These changes involve both:
 an increase in the degree of sorting & a decrease in matrix content.
- Thus an index of the degree of physical maturation might be:

the ratio of grains to matrix.



• Total clay content is a useful index of textural maturation.

Both physical and chemical maturation occur

during the history of a sand population, but they

are not closely related (Johnson and Basu, 1993).

• Thus a chemically mature sand may be

physically immature and vice versa !!!!

it would appear that the most appropriate triangular *classification* to adopt would have *stable* grains at one apex, matrix at the second and unstable grains at MATRIX the third.



- These concepts of maturity can be used as a basis for S.st. nomenclature.
- The total amount of clay in a sand is the best indicator of matrix content.
- *Feldspar* is a common abundant unstable mineral (index of chemical immaturity).
- *Quartz* is the index mineral for the apex of chemical stability.



 Using *quartz, feldspar, and clay* as end-members, a sand classification scheme can be drawn up as shown.



sands may be divided into **two broad textural types**, *the arenites* and



- Arenites, with a matrix content of < 15%, are texturally mature.
- Wackes, with a matrix content of 15 -75%, are texturally immature.
- *Rocks* with >75 % matrix
 are not sandstones but *mudrocks*.

- Similarly, sands can be divided into chemically mature arenites & chemically mature wackes.
- Both have < 25 % of feldspar. These two sandstones are: protoquartzite & quartz-wacke.
- The mature arenites should be called *quartz-arenites*. Mudrocks
- Space is found in the quartz apex for *orthoguartzite sands*,



- Sands with >25 %
 feldspar are
 chemically immature.
- They are divided into:
 arkoses for the
 arenites &
 greywackes for
 the wackes.


Nomenclature and Classification of Sandstones

It is convenient to divide the Sandstones into <u>three groups :</u>

- The quartzites: (including both protoquartzites & orthoquartzites).
- The arkoses
- The wackes: (including both quartz-wacke & greywacke).



Quartzites (Characteristic Features)

- Sands that are mature in texture and mineralogy are broadly referable to as the quartzites or quartz-arenites.
- Among sedimentary petrographers, a quartzite generally refers to a *quartz-rich sand* irrespective of its degree of lithification.
- Quartzites have < 25 % feldspar &
 < 15 % matrix.
- Quartzites are generally divisible into protoquartzites

which contain some feldspars (< 25%) and matrix (< 15%), **& orthoguartzites, which are almost pure silica.**



Quartzites

(Characteristic Features)

- The quartzites are, therefore, sands that are chemically and physically
- *Typical quartzites* are white, pale gray, or pink in color.
- They range from unconsolidated to splintery in their degree of lithification.
- Grain size is variable, but sorting is *generally good* and individual grains are normally *well rounded*.





Mineralogical composition of Quartzites

• The main detrital grains are *quartz* (derived from igneous and metamorphic rocks), and *cherts* (made of quartz, chalcedony, or crystobalite) reworked from sediments.



Fibrous to feathery texture of chalcedony. Unknown formation. Crossed nicols (From Boggs, 2009).

Mineralogical composition of Quartzites

- Rare *feldspar.*
- Mica grains may be present.
- Heavy minerals are generally the stable residue such as zircon, tourmaline, apatite, and garnet.
- Intraformational autochthonous detrital grains are common in quartzites, such as *glauconite*, *phosphate pellets, and skeletal debris.*

Quartzites (Cementation)

- Quartzites possess high porosities and permeabilities at the time of their deposition.
 Why?
- Cementation is generally by calcite or secondary silica.
- But where cement is absent, *quartzites make the best aquifers and hydrocarbon reservoirs of all the sandstone types.*





Origin of Quartzites

- It is probable that most, if not all, quartzites are polycyclic in origin.
- i.e., they have been through more than one cycle of weathering, erosion, transportation, and deposition to achieve the necessary maturity to qualify.
- The majority of quartzites were deposited in marine sand-shoal environments.

Origin of Quartzites

 They also occur in many other environments, including modern deep-sea sands (Hubert, 1964) and ancient turbidites (Sturt, 1961).





ARKOSES (Definition)

- The term arkose was first proposed by *Brogniart* (1826) for a coarse sand *composed of quartz and substantial quantities of feldspar* from the Auvergne in France.
- Arkoses are the product of the incomplete degradation of *acid igneous & metamorphic rocks such as granites and gneisses.*
- Arkose often forms in place on granites forming a transitional weathering zone, termed "granite wash".

Characteristic Features of Arkoses

- The arkoses are sands which are relatively *mature in texture* (i.e. < 15 % clay matrix) & are *immature in mineralogy* (> 25 % feldspar).
- Colour: The typical arkose is a pink or red, less commonly it is gray.
- <u>Pink coloration:</u> is due to the feldspars.
- <u>The red color:</u> owe to absorption of red ferric oxide into the clay matrix.



Characteristic Features of Arkoses

- Arkoses show a wide
 range of grain sizes &
 are often poorly sorted.
- The grains of arkoses are typically *angular to subrounded.*



Characteristic Features of Arkoses

- There is a significant amount of clay matrix.
- For this reason arkoses seldom stay unconsolidated for long time.
- In extreme cases *porosity* can be completely obliterated by a *silica* or *carbonate* cement.

Mineralogy of Arkoses

- The feldspars in arkose are of various types, depending on the nature of the source rock, but microcline & albite more abundant than the less stable calcic feldspars.
- Alteration of the feldspars to sericite & kaolinite is a common feature.





Mineralogy of Arkoses

- Micas are a common accessory minerals.
- Arkoses often contain a *diverse suite of heavy minerals.*
- These may give some *indication of the* source terrain, whether it was *igneous* or *metamorphic*.
- Opaque iron ores are found in addition
 to the stable suite of *zircon, apatite, garnet* & *tourmaline,* as well as many
 other heavy minerals.



Depositional Environments of Arkoses

- Most arkoses seem to occur in alluvium adjacent to granitoid basement.
- Because such sediments are deposited in an oxidizing alluvial fan environment, for many arkoses are termed the *"red beds"* facies
 - assemblage.



Depositional Environments of Arkoses

- Though probably the bulk of arkoses occur in *fluvial environments,* they are also found elsewhere.
- Arkoses mixed with *coralgal debris* occur on modern beaches of the fault-bounded Red Sea.



Characteristic Features of Wackes

- The wackes are texturally *immature* sands with more than 15 % matrix.
- Chemically mature wackes, with less than 25 % feldspar, are termed "quartz-wackes".
- Chemically immature wackes with more than 25 % feldspar, are "greywackes".
- The name "greywacke" comes from the German "grauewacke," which was applied to the Paleozoic sandstones of the Harz Mountain.



Characteristic Features of Wackes

- The **greywackes** are hard, dark graygreen rocks that break with a hackly fracture.
- Greywackes are very poorly sorted with particles ranging from very coarse sand grains down into claygrade matrix.
- They have been described as microconglomerates.
- Grains are commonly *angular* and of *poor sphericity*.





Mineralogical composition of Wackes

- Quartz is the most abundant detrital mineral.
- Feldspar is present, but so also are mafic grains such as horneblende & occasionally, pyroxenes.
- Some of the larger grains are *lithic* rock fragments.
- These may have been derived from volcanics or older metasediments such as quartzite or slate.



Mineralogical composition of Wackes

- Micas are abundant and include both muscovite & biotite, as well as microcrystalline diagenetic chlorite & sericite.
- A diverse suite of *unstable heavy minerals* is also typical.
- All of the detrital grains are set in the **abundant matrix**.
- This is a microcrystalline paste of clay minerals, chlorite, sericite, quartz, carbonate (often siderite), pyrite, and occasionally carbonaceous matter.



Mineralogical composition of Wackes

- The quartz-wackes differ from the typical greywackes in that they lack the diverse suite of unstable detrital minerals.
- In hand specimen quartz-wackes and greywackes are hard to distinguish.
- The poor sorting of these sandstones presupposes that the matrix may be largely syndepositional.



- The typical wackes are largely Pre-Cambrian and Paleozoic in age. This suggests that time, *deep burial*, and/or *high geothermal gradients* are needed to generate greywackes.
- The greywackes are commonly found in pre-Mesozoic flysch facies. These typically occur in *subductive troughs*.
- From petrology and regional setting, it is noted that greywackes are often derived from the rising island arcs of volcanic origin. Hence the unstable suite of mafic minerals and the relatively high percentage of iron and magnesia in greywackes.



- Quartz-wackes are more usually *derived from preexisting sediments.* Sands contribute the quartz, shales produce the clay, and the lithic fraction comes from the indurated equivalents of both.
- Continental quartz-wackes are often red-brown in color due to impregnation of the clay matrix by red ferric oxide.
- With increasing transportation, the quartz-wackes lose some of their clay content and assume a rock type often termed *subgreywacke*.
- This is found in both fluvial and deltaic deposits.

RUDACEOUS ROCKS

موضوع يكلف به مجموعة من الطلاب بالبحث فى مكتبة الكلية وشبكة المعلومات الدولية (الإنترنت) ثم عرضه ومناقشته أثناء المحاضرة



VOLCANICLASTIC ROCKS (PYROCLASTIC ROCKS)



Volcaniclastic sediments

- *Volcaniclastic sediments are classified into two types according to mode of formation :*
- **1. Erosional volcaniclastic sediments:** are produced by normal subaerial or subaqueous processes acting on eruptive rocks.
- *2. Pyroclastic sediments: ejected into atmosphere during volcanic eruptions.*
- > Pyroclasts include:
- **Bombs** that fall close to the vent (larger than 64 mm in diameter).
- Sands which fall around the vent for a distance of kilometers,
- **Dust** which may carried into the upper atmosphere & transported around the world.





VOLCANICLASTIC ROCKS

- Volcaniclastic sediments can be classified into three groups according to their particle size:
- 1. Agglomerates: are the counterpart to conglomerates.
- 2. Sand-grade volcaniclastic sediments.
- 3. Fine-grained volcanic ash.



Agglomerates: are the counterpart to conglomerates.







 Agglomerates are formed by explosive eruptions & by scree movement of volcanic detritus within a caldera & on the flanks of volcanoes.

Sand-grade Volcaniclastics

- Volcaniclastic sands are generally referred to as "tuffs" or "ashes".
- * They may be subaerial or subaqueous.
- Volcaniclastic sands are composed of crystals, glass & rock fragments.
- The crystals: are of minerals associated with the eruption, such as olivine & quartz.
- Glass: occurs as globules & angular irregularly shaped shards.
- Rock fragments: are composite grains of volcanic minerals and glass.





Sand-grade Volcaniclastics

- Volcaniclastic sands are generally poorly sorted.
- Eolian volcaniclastic sands are an exception to this rule.



Sand-grade Volcaniclastics

- Volcaniclastic sands may have good primary porosity & permeability, especially if deposited on beaches or dunes.
- They lose porosity on burial very quickly, because of their unstable mineralogy.
- During shallow burial they undergo hydration & carbonation leading to the formation of authigenic carbonate, clay, laumontite & other zeolites.
- Further burial results in dehydration of clays, albitization of feldspars & development of additional zeolites.



VOLCANIC




VOLCANIC ASH

- Fine-grained volcanic ash tends to undergo intensive post-depositional alteration giving rise to the *bentonites*.
- Palagonite: is a mineraloid produced by the alteration of basalt glass.
- Traces of volcanic ash are a common component of modern pelagic sediments.





Volcaniclastic rocks

- Volcaniclastic rocks are generally poor quality aquifers or petroleum reservoirs.
- They are of interest in mineral exploration.
- They appear to be the primary source of some metalliferous deposits.

AUTOCHTHONOUS SEDIMENTS

AUTOCHTHONOUS SEDIMENTS (INTRODUCTION)

- The second great group of sedimentary rocks is variously referred to as the *chemical group* or *autochthonous sediments group*.
- The chemical rocks are sometimes divided into
 ✓ Organic group (e.g. Carbonate skeletal sands) &
 - ✓ Inorganic group (evaporites).

Major Types of Chemical Rocks

Carbonates	Limestone & Dolomite (Dolostone)	
Evaporites	Anhydrite/Gypsum Halite/Rock salt Potash salts, etc.	
Siliceous rocks	Chert, Radiolarite, Novaculite	
Carbonaceous rocks	Humic group - Coal series Sapropelitic group- Oil shales & Cannel coals	
Sedimentary ironstones	Goethite, Hematite & chamosite	
Phosphates	Collophane, Dahllite, Francolite & Fluorapatite.	

AUTOCHTHONOUS SEDIMENTS (INTRODUCTION)

- This group consists of rocks that form within a depositional basin.
- Not all chemical sediments are syndepositional.
- Diagenetic processes are important in the genesis of some evaporites, dolomites, cherts, ironstones, and phosphates.

AUTOCHTHONOUS SEDIMENTS (Mode of formation)

- 1. The carbonate rocks form by:
 - organic processes,
 - > Direct inorganic precipitation &
 - > Diagenesis.
- 2. Evaporites form by:
 - Inorganic crystallization &
 - > Diagenesis.
- 3. The sedimentary ironstones also form as:
 - Precipitates &
 - > Diagenesis.

AUTOCHTHONOUS SEDIMENTS (Mode of formation)

- 4. Phosphates form largely during early diagenesis in sediment, beneath the sediment/water interface.
- 5. Coal is formed entirely by biochemical processes.
- 6. Hydrous silica (opal), occurs in Tertiary rocks, but appears to dehydrate on burial to silica.





- Carbonates are important aquifers and hydrocarbon reservoirs because of the high porosity which they sometimes contain.
- Carbonate rocks are *intrabasinal in origin*.
- They are deposited at or close to their point of origin.
- Most carbonate rocks are *organic in origin*.
- They are easily weathered and *their weathering products* are transported as solutes.
- They contain a wide spectrum of particle sizes, ranging from whole shells to lime mud of diverse origin.



- Carbonate sediments are deposited with a high primary porosity.
- The carbonate minerals are *chemically unstable*.
- The combination of high primary porosity and permeability, coupled with chemical instability, is responsible for the complicated diagenesis of carbonate rocks.

- Calcium carbonate (CaCO3) is the dominant constituent of modern carbonates and ancient limestones.
- It occurs as two minerals, aragonite and calcite.
- Aragonite crystallizes in the orthorhombic crystal System,
- Calcite crystallizes in the rhombohedral class (Trigonal system).





- Calcite is the more stable variety, while aragonite is unstable.
- A distinction is made between high- and low-magnesium calcite, with the boundary being set at 10 mol%.
- Ancient limestones are composed largely of *low-magnesium calcite*, while modern carbonate sediments are made mainly of *aragonite & high-magnesian calcite*.
- Skeletal aragonite & calcite also contain minor amounts of strontium, iron, and other trace elements.

- Dolomite is another important carbonate mineral, giving its name also to the rock.
- Dolomite is calcium magnesium carbonate [CaMg(CO3)2].





- Dolomite does not occur as a biogenic skeletal mineral.
- Dolomite is generally found either crystalline, as an obvious secondary replacement of other carbonates, or rarely as a primary or penecontemporaneous replacement mineral in cryptocrystalline form.

 Isomorphous substitution of some magnesium for iron is found in the mineral termed <u>ferroan dolomite</u> or <u>ankerite</u> [Ca(Mg,Fe)(CO3)2]



- *Siderite (FeCO3)* is one of the rarer carbonate minerals.
- *It occurs,* as a primary precipitate, **in ooliths.**



MAIN COMPONENTS OF CARBONATE ROCKS

- Carbonate rocks have four main components: grains, matrix, cement, & pores.
- The grains of carbonate
 rocks, commonly mono *minerallic*, are texturally
 diverse and polygenetic.



MAIN COMPONENTS OF CARBONATE ROCKS

	Main Components of Carbon	ts of Carbonate Rock		
I. Casian	 (a) Detrital grains (b) Skeletal grains (c) Peloids (including fecal pellets) 	{ Lithoclasts Intraclasts		
I. Grains	(d) Lumps	Composite grains Algal lumps		
	(e) Coated grains	{ Pisoliths Algally encrusted grains		
II. Matrix	{ Micrite Clay	5, 5		
III. Cement IV. Pores	Sparite			

GRAINS

- *Grains* are the particles that support the framework of a sediment.
- They are generally of *sand grade or larger*.
- carbonate grains are of many types as listed in the following table:

GRAINS (LITHOCLASTS)

The detrital grains are of two types, *Lithoclasts & intraclasts.*

1) Lithoclasts (Rock fragments):

These are grains of *non-carbonate material* which originated **outside** the depositional basin.

 Quartz grains are a typical example of lithoclasts and, as they increase in abundance, limestones grade into sandy limestones & thence to calcareous sandstones.



Well-rounded detrital quartz sand grains are scattered throughout dolomitized carbonate mudstone.



GRAINS (INTRACLASTS)

- 2) Intraclasts: These are fragments of reworked carbonate rock which originated within the depositional basin.
- Early cementation followed by penecontemporaneous erosion is responsible for the generation of intraclasts.

Erosion of weakly cemented substrate





- *Skeletal detritus* are the most important of all grain types.
- They are termed *bioclasts*.
- They are composed of *aragonite or calcite* with varying amounts of trace elements.
- The crystal habit of skeletal matter is varied, ranging from *acicular aragonite crystals of lamellibranch shells to single calcite crystals of echinoid plates.*
- The size of skeletal particles is very variable, ranging from the largest shell to individual microscopic crystals.

Skeletal Grains in Carbonate Rocks



Planktonic Foraminifera, PPL



Bryozoan colony, PPL, X36



Bivalve, PPL, X25



Echinoid spine, PPL, X50

Skeletal Grains in Carbonate Rocks



A longitudinal axial section of a single nummulite



A high-magnification view shows the distinctive oval outlines of minute coccoliths.



An echinoderm fragment with characteristic single-crystal and uniform "honeycomb" microtexture (small pores filled with micrite). Table: Shows the mineralogy of the major groups of carbonate-secreting organisms,after Scholle (1978).

Mineralogy of Skeletal Organisms					
Taxon	Aragonite	Calcite %Mg 0 5 10 15 20 30 35	Both Aragonite & Calcite		
Calcareous Algae:					
Red	R				
Green	•				
Coccoliths		•			
Foraminifera:					
Benthic	R	······································			
Planktonic		••			
Sponges:	R				
Coelenterates:					
Stromatoporoids		•?			
Milleporoids	•				
Rugose		•?			
Tabulate		•?			
Scleractinian	•				
Alcyonarian	R	••			
Bryozoans:	R	• •	R		
Brachiopods:		••			
Molluscs:					
Chitons	•				
Bivalves			•		
Gastropods		0-0			
Pteropods	•				
Cephalopods					
Belemnoids & Aptychi		•			
Serpulids:	•	• •	•		
Arthropods:					
Decapods		••			
Ostracodes			all she are the		
Barnacles		0-0			
Trilobites					
Echinoderms:		•••••••••••			
Common R Rare					

GRAINS (PELOIDS)

- Peloids are a third major grain type.
- These were first defined as structureless cryptocrystalline carbonate grains of some 20-60/µm in diameter (McKee and Gutschick, 1969).
- Peloids form in many different ways:

Pellets are peloids of fecal origin, excreted by marine invertebrates.



GRAINS (PELOIDS)

- Pelletoids are peloids formed by the micritization of skeletal grains through the action of endolithic algae.
- Peloidal deposits are especially characteristic of lagoons and other sheltered shallow innershelf environments.



NOMENCLATURE AND CLASSIFICATION OF CARBONATES

NOMENCLATURE AND CLASSIFICATION OF CARBONATES

- So many parameters can be used to define
 - carbonate rock types.
- These include chemical composition, grain size,
- particle type, type and amount of porosity, degree of
 - crystallinity, quantity of mud, and so on.
- Folk (1962) & Dunham (1962) proposed a series of
 - terms and groupings which are widely used today.

- Folk's classification is applicable to thin-section analysis.
- It is not an easy classification to use in the field.
- Folk (1962) divided limestones *into <u>four</u> main classes as follows:*
- <u>The first two classes</u> include rocks composed largely of carbonate grains (allochems); these are termed the "allochemical limestones".
 - One class is dominated by sparite cement.
 - > The other is dominated by *micrite matrix*.

- The third class is for rocks *lacking grains*, termed the *"orthochemical limestones"*.
 - > This group includes *micrite lime mud carbonates*.
- The fourth group is for rocks made of in-place (in situ) skeletal fabrics.
 - This group, the autochthonous reef rocks, includes *"coral biolithite" & "algal stromatolites".*



A 15-cm-wide v. section of laminated algal stromatolite

Stained thin section of stromatolite, PPL, x 9

Polished hand specimen of stromatolite Magnification, x1.8





- Folk proposed a "*bipartite nomenclature*" for the allochemical rocks (the first two classes).
 - The first (prefix) defined the *grain type* &
 - The second (suffix) denoted whether *sparite* or

micrite predominated.

For example: "oosparite" ??

& "pelmicrite" ??

 Where more than one allochem type was present, two should be used, with the major one first, as in "biooosparite".

- When both *micrite & sparite* are present, they could both be compounded with the dominant constituent first: as in

DUNHAM's CLASSIFICATION (1962)

- Dunham (1962) focused upon depositional limestone textures rather than upon the identity
 - of specific kinds of carbonate grains.
- He placed, like Folk the *in-place reef rocks* in a separate class, termed them *boundstones*.
- Dunham divided the rest of the carbonates into *four groups* according to whether their

DUNHAM's CLASSIFICATION Of carbonates (1962)

Depositional texture recognizable				Depositional texture not recognizable	
Components not bound together during deposition Components					
Contains carbonate mud (clay / fine silt)		Lacks mud and is	were bound together during denosition		
Mud su	oported	Grain	supported	deposition	Constalling
Less than 10% grains	More than 10% grains	Dasketone	Crainatana	Poundatana	
5 mm	5 mm	Packstone 5 mm	Grainstone 5 mm	soundstone	Crystamile 5 mm
- Grainstones are grain-supported sands with no micrite matrix.
- Packstones are grain-supported sands with minor amounts of matrix.
- Wackestones are mud-supported rocks with dispersed number of grains.
- *Mudstones* are carbonate muds.

- The nomenclature is simple & identification can be made with only a hand lens.
- Furthermore, this nomenclature, by drawing attention to *fabric and matrix content*, gives an index of the depositional energy.
 - Thus the mud-supported limestones may indicate deposition in a low-energy environment.
 - By contrast, the matrix-free grain-supported rocks suggest deposition in high-energy environment in which no mud could come to rest.

- The textural maturity of carbonate sediments may be considered is analogous to the maturity of terrigenous sands.
 - A mature carbonate sediment is one that is composed exclusively of one grain type.
 - The different carbonate grains result from different processes operating in different environments.
 - Thus a carbonate rock composed of a *single grain type* indicates genesis from a *single process.*
 - Carbonate rock composed of *two grain types* reflects genesis from *two processes*, and so forth (Smosna, 1987).

The grain size, sorting, and matrix content can be used with reservations as indicators of hydro-dynamic environment in carbonate rocks.

The origin of micrite is polygenetic.

 Micrite may develop by *bioturbation, algal micritization, cementation* & by infiltration due to high permeability.

FOLK's & DUNHAM's CLASSIFICATION Of carbonates

 Finally, the classifications and nomenclature of carbonates proposed by *Folk and Dunham* are extremely useful and, used in conjunction, encompass most varieties of limestones with flexibility and finesse.

Nomenclature and classification of limestones according to schemes of Dunham (1962), and Folk (1962).

	GRAIN TYPES				
MUDSTONE < 10% grains	Lime mud, micrite, calcilutite, chalk				
1	PELLETS	SHELL DEBRIS	OOLITHS	INTRACLASTS	
WACKESTONE > 10% grains, mud supported					
	Pelmicrite	Biomicrite	Oomicrite	Intromicrite	
PACKSTONE > 5% mud, grain supported					
	Pelmicsparite	Biomicsparite	Oomicsparite	Intromicsparite	
GRAINSTONE < 5% mud					
	Pelsparite	Biosparite	Oosporite	Intrasparite	
BOUNDSTONE original components bound together	Reef rock , biolithite				

CLASSIFICATION Of carbonates

- Sadly neither of both schemes embrace the most important aspect of carbonates for many geologists, namely, their reservoir characteristics, in terms of porosity and permeability.
- Archie (1952) classified carbonate reservoirs based on the concept that *pore-size distribution controls permeability and petroleum saturation,* and that *pore-size distribution is related to the fabric of the rock.*

CLASSIFICATION Of carbonates (Lucia, 1995 & 1999)

- Lucia (1995 & 1999) developed scheme classifies carbonates (limestones & dolomites) into three petrophysical classes according to their pore-type:
- Class (1): Interparticle (intergranular) vugs, Class (2): Separated (isolated) vugs, Class (3): Communicating (connected) vugs.
- These three petrophysical classes may then be subdivided acording to Dunham's terminology of *grainstone, packstone, wackestone & mudstone.*
- In general Lucia's classification follows the Dunham carbonate classification, except that the packstone fabric group has been divided into grain-dominated and mud-dominated classes.

CLASSIFICATION Of carbonates (Lucia, 1995 & 1999)



- Crystallization: means the infilling of primary interand intraparticle porosity by the drusy growth of sparite out from the pore walls.
 - > This naturally results in a **decrease in porosity.**



Stained thin section, PPL, X 122

- Neomorphism: is the term applied to describe the recrystallization or replacement of a mineral (Folk, 1965).
- Neomorphism can lead to both increasing, or unaltered porosities.
- Recrystallization defined as neomorphism in which the mineralogy is unchanged, does not significantly alter the amount or type of

- 2) polymorphic transformations, in which one mineral replaces another, can have large effects on rock porosity.
 - i. One of the earliest diagenetic changes is the *transformation of aragonite to calcite.*
 - > This results in *an increase* in total rock volume *of 8%,*
 - i.e. decrease in rock porosity of 8%.
 - ii. The replacement of *calcite by dolomite (dolomitization)* can cause an overall *contraction of the rock by as much as 13%* (Chilingar and Terry, 1964) , *i.e. increase in rock porosity of 13%.*

The intercrystalline porosity caused

this replacement

that makes secondary dolomites

such attractive reservoir rocks.





Conversely, a decrease in porosity is caused by the transformation of dolomite to calcite; (dedolomitization or calcitization) as it is called (Shearman et al., 1961).

- Leaching: is an important process giving rise to secondary porosity.
- Solution porosity: may be due to the selective solution of matrix, cement, or specific grain types.
- Vuggy porosity: results from the solution of pores whose boundaries cross-cut the fabric.
- Moldic porosity: describes the selective solution of one particular grain type.
- for example porosity.





- *Silicification:* is another characteristic diagenetic process in carbonate rocks.
- The silicification of lime muds is not of petrophysical significance.
- In calcarenites and reef rocks, silicification can be an important destroyer of primary porosity when it develops as a chalcedonic

cement.





INTRODUCTION

- The evaporites form by the evaporation of salt-rich fluids.
- Evaporites can form in both marine and nonmarine environments.
- Owing to differences in the dissolved mineral composition of the feed waters, different suites of minerals tend to form in the two environments.
- The evaporites include some common minerals (*Eight* minerals as mention in the following table), but the total number of evaporite minerals is vast.
- Approximately 80 minerals in total have been reported from evaporite deposits *(Stewart, 1963)*.

COMMON EVAPORITE MINERALS

Some of the More Common Evaporite Minerals

Name	Composition	
Anhydrite Gypsum Polyhalite Epsomite	$\left. \begin{array}{c} CaSO_{4} \\ CaSO_{4} \cdot 2H_{2}O \\ CaSO_{4} \cdot MgSO_{4} \cdot K_{2}SO_{4} \cdot 2H_{2}O \\ MgSO_{4} \cdot nH_{2}O \end{array} \right\}$	Sulfates
Halite Sylvite Carnallite Bischofite	$\left. \begin{array}{l} NaCl \\ KCl \\ KMgCl_3 \cdot 6H_2O \\ MgCl_2 \cdot 6H_2O \end{array} \right\}$	Chlorides

Major Evaporite Minerals (cited from Boggs,

Mineral	Formula	Mineral	Formula
Anhydrite	CaSO ₄	Leonhardtite	MgSO ₄ · 4H ₂ O
Antarcticite	CaCl ₂ · 6H ₂ O	Leonite	MgSO4 · K2SO4 · 4H2O
Aphthitalite (glaserite)	$K_2SO_4 \cdot (Na,K)SO_4$	Loewite	$2MgSO_4 \cdot 2Na_2SO_4 \cdot 5H_2O$
Aragonite **	CaCO ₃	Mg-calcite**	$(Mg_xCa_{1-x})CO_3$
Bassanite	CaSO ₄ · 1/2H ₂ O	Magnesite**	MgCO ₃
Bischofite	MgCl ₂ · 6H ₂ O	Meyerhoffite	Ca2B5O11 · 7H2O
Bloedite (astrakanite)	$Na_2SO_4 \cdot MgSO_4 \cdot 4H_2O$	Mirabilite	$Na_2SO_4 \cdot 10H_2O$
Borax (tincal)	Na ₂ B ₄ O ₇ · 10H ₂ O	Nahcolite	NaHCO ₃
Boracite	Mg ₃ B ₇ O ₁₃ · Cl	Natron	Na ₂ CO ₃ ·10H ₂ O
Burkeite	Na ₂ CO ₃ ·2Na ₂ SO ₄	Nitratite (soda nitre)	NaNO ₃
Calcite**	CaCO ₃	Nitre (salt petre)	KNO3
Carnallite	MgCl ₂ ·KCl·6H ₂ O	Pentahydrite	MgSO4 · 5H2O
Colemanite	Ca2B5O11 · 5H2O	Pirssonite	CaCO3 · Na2CO3 · 2H2O
Darapskite	NaSO4 · NaNO3 · H2O	Polyhalite	2CaSO4 · MgSO4 · K2SO4 · H2O
Dolomite**	$Ca_{(1+x)}Mg_{(1-x)}(CO_3)_2$	Proberite	NaCaB ₅ O ₉ · 5H ₂ O
Epsomite	MgSO ₄ · 7H ₂ O	Priceite (pandermite)	$CaB_4O_{10} \cdot 7H_2O$
Ferronatrite	3NaSO4 · Fe2(SO4)3 · 6H2O	Rinneite	FeCl ₂ · NaCl · 3KCl
Gavlussite	CaCO ₂ ·Na ₂ CO ₂ ·5H ₂ O	Sanderite	MgSO4 · 2H-O
Glauberite	CaSO ₄ · Na ₂ SO ₄	Schoenite (picromerite)	MgSO ₄ · K ₂ SO ₄ · 6H ₂ O
Gypsum	CaSO ₄ · 2H ₂ O	Shortite	2CaCO ₃ · Na ₂ CO ₃
Halite	NaCl	Sylvite	KCl
Hanksite	9Na2SO4 · 2Na2CO3 · KCl	Syngenite	CaSO4 · K2SO4 · H2O
Hexahydrite	MgSO ₄ · 6H ₂ O	Tachyhydrite	CaCl ₂ · 2MgCl ₂ · 12H ₂ O
Howlite	H ₅ Ca ₂ SiB ₅ O ₁₄	Thenardite	Na ₂ SO ₄
Ikaite**	CaCO ₃ · 6H ₂ O	Thermonatrite	NaCO ₃ ·H ₂ O
Inyoite	Ca2B6O11 · 13H2O	Tincalconite	$Na_2B_4O_7 \cdot 5H_2O$
Kainite	4MgSO4 · 4KCl · 11H2O	Trona	NaHCO3 · Na2CO3 · 2H2O
Kernite	Na2B4O7 · 4H2O	Tychite	2MgCO3 · 2NaCO3 · Na2SO4
Kieserite	MgSO ₄ · H ₂ O	Ulexite	NaCaB ₅ O ₉ · 5H ₂ O
Langbeinite	2MgSO4 · K2SO4	Van'thoffite	MgSO ₄ · 3Na ₂ SO ₄

- Previously, it was widely accepted that evaporites formed largely by the *precipitation or crystallization of salts at the sediment : water interface* (e.g., Borchert and Muir, 1964).
- It is now largely accepted that the genesis of evaporites by diagenesis is the rule (Kirkland and Evans, 1973; Kendall, 1984; Sonnenfeld, 1984).

- It appears most probable that evaporites form from saline-rich fluids (brines).
- Brines may be generated by concentration of seawater, by evaporation or freezing, or as residual connate fluids in the subsurface.
- Secondary brines can form where meteoric groundwater passes through and dissolves previously formed evaporites.
- Normal ocean water contains 3.45 % by weight of *dissolved* substances.
- 99.9% of the dissolved material comprises *nine ions*.
- These ions are: Sodium, Magnesium, Calcium, Potassium, Strontium, Chloride, Sulphate, Bicarbonate and Bromine.

- Some of the earliest works on the genesis of evaporites was to study salts formed from the evaporation of seawater (Usiglio, 1849; Van't Hoff and Weigert, 1901).
- These studies demonstrated two main facts; that
 - 1. Inconceivable quantities of seawater were necessary to form observed volumes of evaporites in a closed system.
 - The observed percentages of salts in an evaporite assemblage differ somewhat from those produced by the evaporation of seawater.

 The explanation that evaporite formation characteristically occur in basins that had restricted access to the sea (silled basin).





- In a restricted basin the seawater flow from the open ocean into the basin by underground seepage.
- Here excessive evaporation concentrates the seawater.
- The incipient brine sinks to the basin floor because of its higher density.



- The sill prevents drainage of the brine out to the open sea.
- Continuous recycling of the brine increases concentration to the point at which evaporites begin to crystallize on the basin floor.





- This process would be aided by the fluctuating sea level, which allows repeated influxes of water over the sill,
- followed by a drop in water level so as to completely restrict the body of brine.



- This is the classic *"evaporating dish mechanism"* for evaporite genesis (Sloss, 1969).
- Supporting evidence for this mechanism includes the fact that evaporites tend to be zonally arranged within a basin, with salts requiring higher salinity for their formation occurring toward the depocenter.

- Similarly, evaporite minerals tend to be cyclically arranged in the same motif (as shown below).
- This cyclicity is classically demonstrated in the *Zechstein evaporites of the North Sea basin* and also found in most other examples.

	Potassium salts (carnallite, polyhalite, etc.)		
Increasing salinity	Rocksalt (halite)	Brine	
	Anhydrite	DIIIC	
	Dolomite		
	Limestone	Normal seawater	

- Evaporite minerals are of great economic importance for three reasons,
 - 1. They are an economic material in their own right.
 - 2. They are closely related to the genesis and entrapment of hydrocarbons, and
 - 3. There is a strong presumption that evaporite associated brines play an important role in the genesis of certain metallic ores (Melvin, 1991).

- Evaporites are a natural resource of great importance, that *supply a large proportion of the world's requirements for the rare earth elements.*
- Evaporites are of importance in the search *for oil and natural gas for three reasons:* source, structure and seal (Buzzalini et al., 1969).
- The crests of salt domes *develop a diagenetic cap rock* of limestone, dolomite, anhydrite, gypsum, native sulfur, and diverse sulfide minerals (Kyle and Posey, 1991).

Salt domes host a series of potential hydrocarbon traps, both domal anticlines above the cap rock and faulted flank traps.





Various types of trap are shown as follows: (A) Weathered basement on fault block, (B) anticlinal trap, (C) limestone reef, (D) truncation trap, (E) closure over salt dome, (F) trap due to sealing fault.

Finally, evaporites are significant because

they provide an ideal reservoir seal, combining a maximum of plasticity w

a minimum of permeability



INTRODUCTION

- Phosphorus is an essential element of all living matter, both plant and animal.
- Phosphate minerals are extensively used *as agricultural fertilizers*.
- Some phosphates occur in igneous rocks as apatite.
- The majority of economic phosphates are in sedimentary rocks.
- Phosphorus-rich sedimentary rocks are called by a variety of names: *phosphate rock, phosphates, phosphatites, and phosphorites.*

MINERALOGY OF PHOSPHATES

- Typical phosphate minerals are admixtures of the phosphate radical (PO₄) with calcium, water, and traces of fluoride and uranium.
- Sedimentary phosphates are composed of phosphate minerals, *all of which are varieties of apatite.*
- The principal varieties are:
 ➤ Fluorapatite [Ca₅(PO₄)₃F],
 ➤ Chlorapatite [Ca₅(PO₄)₃Cl], and
 ➤ hydroxyapatite [Ca₅(PO₄)₃OH].

MINERALOGY OF PHOSPHATES

- Most sedimentary phosphates are *carbonate fluorapatites* in which up to *10 % carbonate ions* can be substituted for phosphate ions to yield the general formula [*Ca₁₀ (PO₄, CO₃)₆ F₂₋₃*]
- These carbonate fluorapatites are commonly called *francolite*.
- *Francolite* is not a formal mineral name; some workers prefer to use the name *carbonate-fluorapatite (CAF) instead*.
- Although *more than 300 phosphate minerals* are known, *francolite* is virtually the only phosphate mineral that *occurs in unweathered marine phosphorites.*

MINERALOGY OF PHOSPHATES

- The term *collophane* is often used for sedimentary apatites for which the exact chemical composition has not been determined.
- Phosphorites commonly also contain some *detrital quartz and authigenic chert*.
- Both *calcite and dolomite* may occur in phosphorites, and dolomite may be particularly abundant.
- Glauconite, illite, montmorillonite and zeolites may also be present in some deposits.
- **Organic matter** is a characteristic constituent of many phosphorites (Nathan, 1984).
OCCURRENCE OF PHOSPHATES

- Most phosphate deposits occur in marine sedimentary sequences.
- They have been reported in rocks of all ages and on all continents.
- Phosphates occur in sedimentary rocks as matrix and as nodules, ooliths, pellets and phosphatized shells, bones, teeth, and coprolites.
- They also occur as *a replacement of limestones* (Bentor, 1980).

- Seawater is generally saturated with *phosphate ions*, ranging from *0 to 3 ppm PO*₄ *in deep cold water* to about *0.01 ppm in warm surface water*.
- The solubility of phosphate decreases with increasing temperature and increasing pH.
- These changes occur, and phosphates thus tend to be precipitated, where deep cold oceanic water wells up into shallower warmer waters.

- The most significant locus at the present time appears to be *along the western coasts of South America and Africa.*
- Phosphate removed from the seawater by organisms returns again when they die, and settles on the sea bed within miscellaneous organic matter.
- Phosphates become concentrated during early compaction of the mud.

- Constant agitation winnows out the lighter material to leave denser incipient phosphate mud pellets.
- These continue to become enriched with phosphate, as do *bones, teeth, and shell debris*.
- In this manner, bedded phosphate rock (phosphorite) is formed.
- The optimum depth for phosphate formation *ranges from 30 to 200m.*

- Eight factors favor phosphate formation (Brown, 1994):
 - 1. A broad shallow shelf,
 - 2. An adjacent major ocean,
 - 3. A low latitude (<40°)
 - 4. High organic productivity,
 - 5. Shallow marine sedimentation,
 - 6. Minimal terrigenous input,
 - 7. A marine transgression, and
 - 8. A suitable environmental trap, such as a bay, estuary or carbonate bank.

STRATIGRAPHIC CHARACTERISTICS OF PHOSPHORITES

- In ancient phosphorite deposits, phosphate-rich layers typically occur interbedded with carbonate rocks, mudrocks, or chert.
- A characteristic feature of many major phosphorite accumulations is the triple association of phosphate, chert, and sediments containing abundant organic carbon.

- Phosphorites have textures that resemble those in limestones. Thus, they may contain ooids, peloids, fossils (bioclasts), and clasts.
- These grains are composed of apatite.
- Some phosphorites lack distinctive granular textures and are composed of fine, micrite-like, textureless collophane.
- Cook (1976) refers to this kind of phosphorite as phospholutite.

- Most grains in phosphorites are sand size; however, particles >2 mm may occur.
- These large grains are commonly referred to as *nodules* and can range in size from 2 mm to several tens of centimeters.
- No scheme for naming and classifying of phosphorites on the basis of texture appears to be in wide use at this time (Boggs, 2009).

- Because phosphorites have textures similar to those in limestones (e.g. peloidal, oolitic, fossiliferous), some geologists have suggested using modified limestone classifications to distinguish different kinds of phosphorites.
- Slansky (1986) used a classification system based on Folk's (1962) limestone classification.

- *Cook and Shergold (1986) and Trappe (2001)* suggested adopting Dunham's (1962) carbonate classification.
- Use of these modified carbonate classification schemes yields names such as *wackestone phosphorite (Cook and Shergold) and phosclast wackestone (Trappe).*

KINDS OF PHOSPHATE DEPOSITS

- Phosphorite deposits can be divided into *five principal groups* on the basis of bedding characteristics and the principal types of phosphate materials that make up the deposits as follows:
 - Bedded phosphorites: form distinct beds of variable thickness, commonly interbedded with carbonaceous mudrocks, cherts, and carbonate rocks.
 - The phosphorite occurs as peloids, ooids, pisoids, phosphatized fossils and skeletal fragments, and cements.

KINDS OF PHOSPHATE DEPOSITS

- 2. Bioclastic phosphorites: are a special type of bedded phosphate deposit composed largely of vertebrate skeletal fragments such as fish bones, shark teeth, fish scales, coprolites, etc.
- Nodular phosphates: are spherical to irregularshaped nodules ranging in size from a few centimeters to a meter or more.
 - Phosphatic grains, pellets, shark teeth, and other fossils may occur within the nodules.

KINDS OF PHOSPHATE DEPOSITS

- 4. Pebble-bed phosphorites: are composed of phosphatic nodules, phosphatized limestone fragments, or phosphatic fossils that have been mechanically concentrated by reworking of earlier-formed phosphate deposits.
- 5. Guano deposits: are composed of the excrement of birds that has been leached to form an insoluble residue of calcium phosphate.

SILICEOUS ROCKS

موضوع يكلف به مجموعة من الطلاب بالبحث في مكتبة الكلية وشبكة المعلومات الدولية (الإنترنت) ثم عرضه ومناقشته أثناء المحاضرة

COAL

موضوع يكلف به مجموعة من الطلاب بالبحث فى مكتبة الكلية وشبكة المعلومات الدولية (الإنترنت) ثم عرضه ومناقشته أثناء المحاضرة

Practical course-Sedimentary rock (ii).

Section No	Contents
1	Handspecimens description.
2	Coarse grains clastic sediments(Conglomerate& Breccia).
3	• Medium grains clastics sediments (sandstone)
4	• Medium grains clastics sediments (sandstone
5	• Fine grains clastic sediments(siltstone, clay or mudstone).
6	Carbonate rocks (limestone).
7	Carbonate rocks (Dolomite).
8	Phosphates.
9	Phosphates.
10	• Evaporites (Gypsum & Rock salts).
11	• Silicious rocks(iron bearing sediments)
12	Revision

Week (1-3)

Item of description:

- Grain size
- Grain shape
- Sorting
- Matrix/cement
- Origin
- Name

Sandstone







Quartz arenite.



Quartz arenite

Week (4):





Quart arenite

Week (5):

litharenite



Arcose



Lithicarenite.

Week (6)



Nummulitic limestone

Non-skeltal component



Ooids



Ooids



Ooids



Ooids



Week (7):



Carbonate rocks



Carbonate rich skeltal



Calcareous cemented skeltal grains



Week (8):

Limestone rich Gastropods



Oolitic grainstone



Dolomite



Dolomitized grainstone

Week (9):



Anhydrite







Iron stone



Week (10):



Bertheriene-chamosite



Phosphorite

Week (11):



Phosphorite



Phosphorite